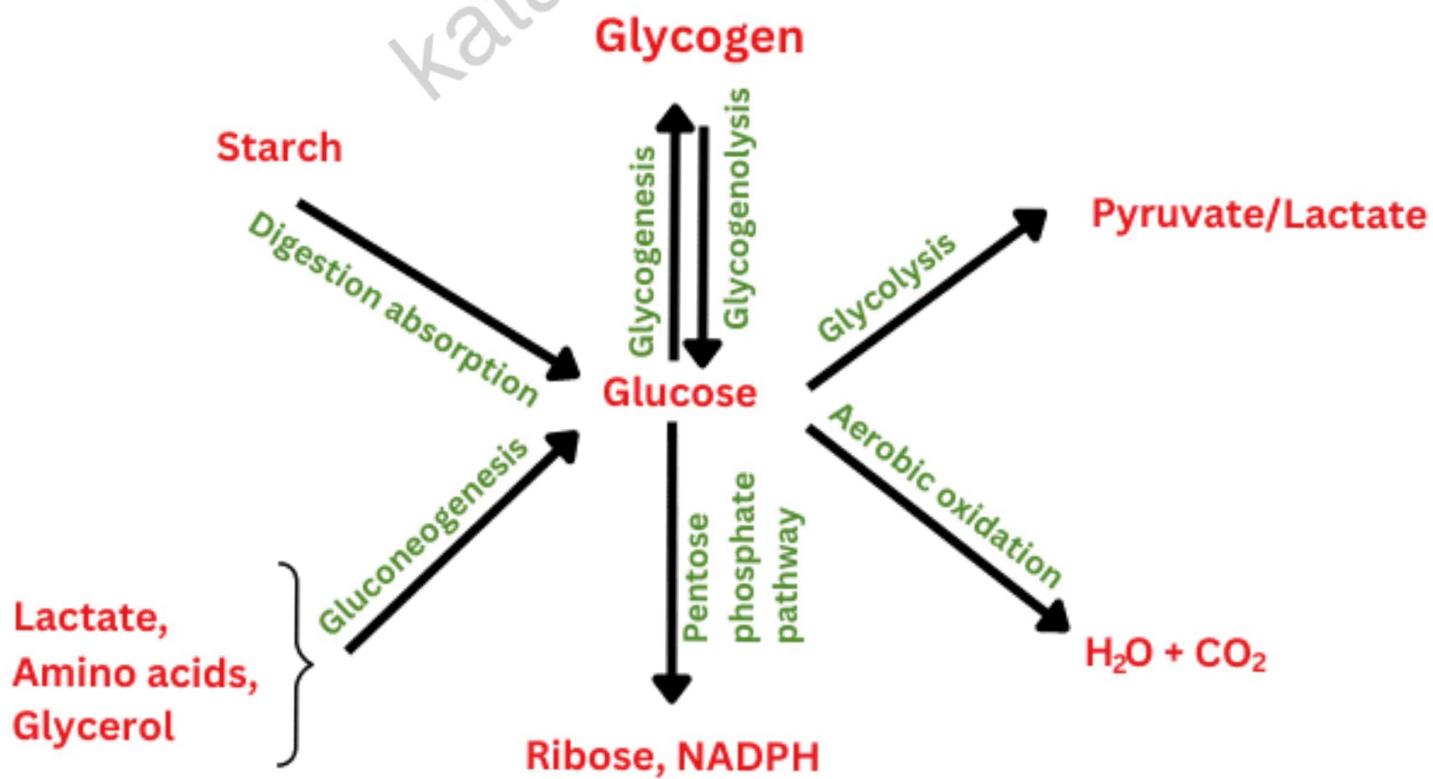


KATARIA PHARMACY
Biochemistry
UNIT-II

* Carbohydrate Metabolism :-

- Biochemical processes involved in Synthesis, breakdown and inter-conversion of Carbohydrates in living organism are collectively called as Carbohydrate Metabolism.
- Glucose is the essential molecule of Carbohydrate Metabolism which participates in various metabolic pathways.
- Insulin is the primary metabolic hormone synthesised in pancreas and regulates glucose level in blood.

Carbohydrate metabolism

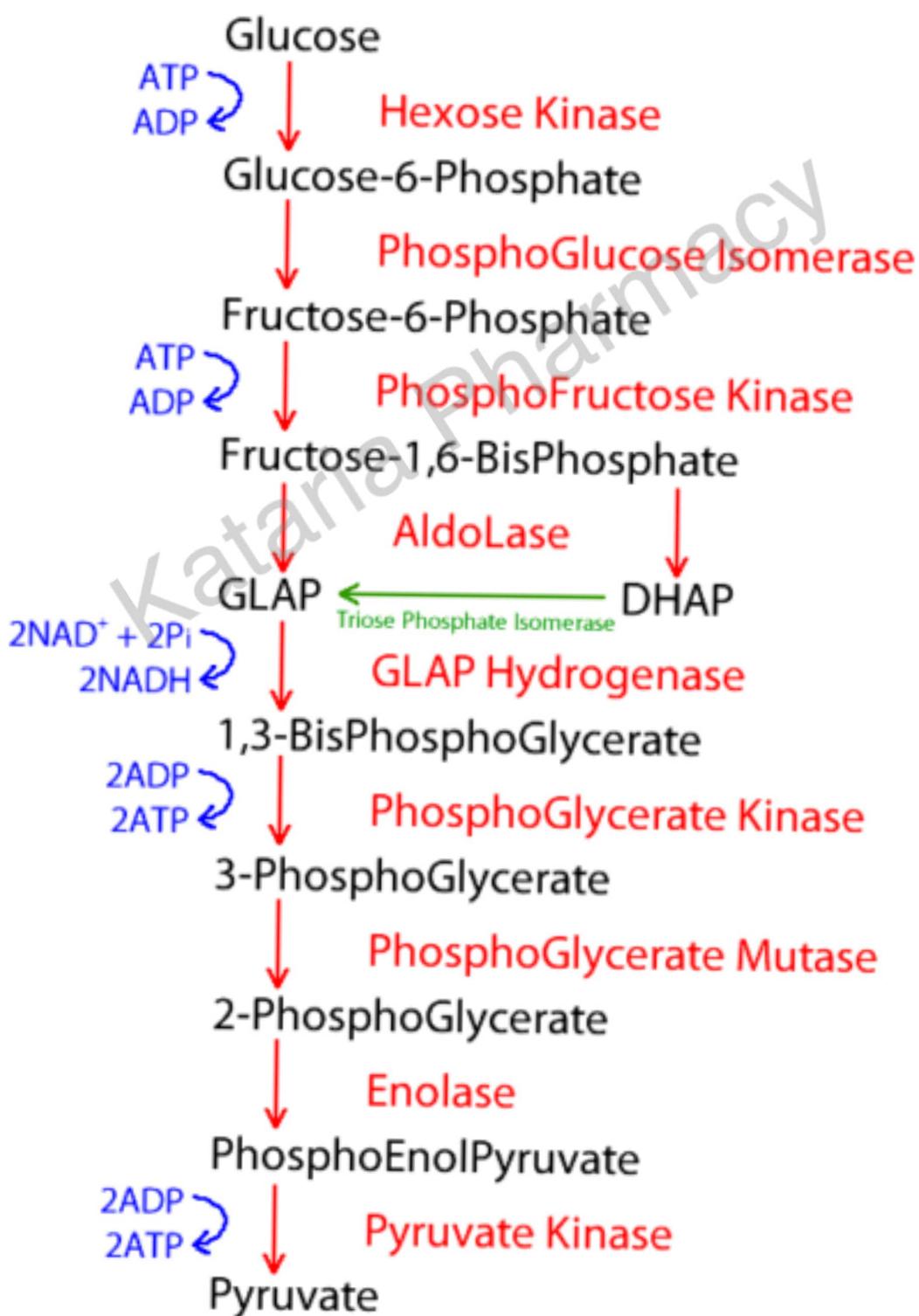


Major Pathways of Carbohydrate Metabolism :-

① Glycolysis :-

Glycolysis is an important pathways of Carbohydrate metabolism and occurs Cytosol of a living cell.

- In glycolysis one mol glucose yield two moles of Pyruvate, ATP and NADH.



Energetics:

- Net gain in glycolysis is 2 ATP, 2 pyruvate and 2 NADH₂ molecule.
- Glycolysis is a low energy process as pyruvate, and NADH₂ molecule captures much of the energy.

Table: Input and output of material during Glycolysis

<u>Total input</u>	<u>Total output</u>
1 Molecule of glucose (6C)	2 molecule of pyruvic acid (2x3C)
2 ATP	4 ATP
4 ADP	2 ADP
2 x NAD	2 x NADH ₂
2 x P _i	2 x H ₂ O

- Total 8 ATP generate during one cycle of glycolysis.

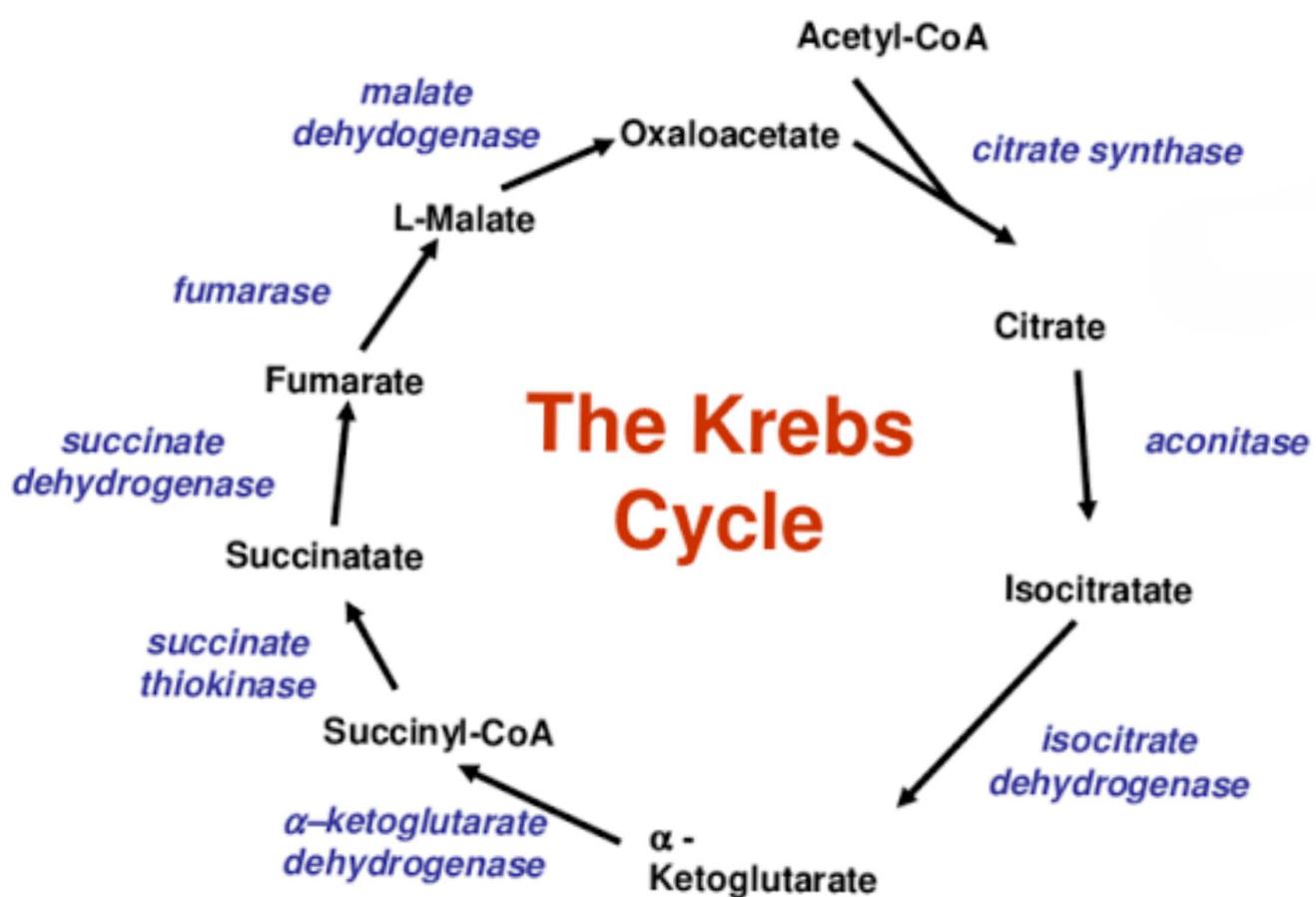
Significance:

- ① In glycolysis pathway, glucose is oxidised to yield pyruvate
- ② Enzyme involved in glycolysis pathway are present in cytosol of a cell.
- ③ All cell utilise glucose for energy
- ④ In a single pathway of glycolysis two molecule of ATP and NADH are generated.
- ⑤ Energy produced by anaerobic glycolysis provides reduce oxygen supply in cell.
- ⑥ Regulation of glycolysis is important to maintain ATP balance.

② Citric acid Cycle:

- The citric acid Cycle (Krebs Cycle or tricarboxylic acid - TCA Cycle) is the most important metabolic pathway for the energy supply to the body.
- ATP is synthesized in Krebs Cycle. Citric acid cycle essentially involves the oxidation of acetyl CoA to CO_2 and H_2O .

Pathway



- Once pyruvate undergoes oxidative decarboxylation to form acetyl CoA, this acetyl CoA then enters the Citric Acid Cycle, also known as the Krebs Cycle.
- The Citric Acid Cycle is a series of eight reactions that break down acetyl CoA, the product of pyruvate decarboxylation, to yield high-energy electron-carrier molecules such as NADH and FADH₂.
- The purpose of the Citric Acid Cycle is to harvest the "high energy" electron from the Carbon fuel source that ultimately come from glucose.

Step 1: Oxaloacetate and acetyl CoA react to form citryl CoA, which then reacts with water to form citrate. The purpose of this step is to create a molecule that will ultimately undergo decarboxylation.

Step 2: The hydroxyl group on citrate is not positioned correctly to undergo a decarboxylation, therefore citrate is transformed into isocitrate by an enzyme called aconitase.

Step 3: This is the first oxidation-reduction reaction creation of the Citric Acid Cycle. Hence we first reduce NAD⁺ into NADH and then produce the α -ketoglutarate via an oxidative decarboxylation reaction.

Step 4: This is the second oxidative decarboxylation reaction we form a four-Carbon Molecule, releasing CO₂ and forming NADH

Step 5 - In this Step, we transfer a phosphate group from Succinyl CoA into a GTP to form GTP. This is the only step in the citric acid cycle that produces with a high phosphoryl transfer potential.

Step 6-8 - The final three steps of the citric acid cycle involve the regeneration of the oxaloacetate molecule.

- In this process of the regeneration of oxaloacetate, also form FADH_2 (step 6), NADH (step 8) and use water (step 7)

Product = 4CO_2 , 6NADH , 2FADH_2 , 2GTP , 4H^+ , 2CoA .

③ HMP Shunt :-

- HMP (Hexose mono phosphate) Shunt is also known as pentose phosphate pathway.
- It is an alternative pathway for glucose oxidation and is a multi-cycle process.
- Enzymes of HMP shunt are located in Cytosol.

Pathway :-

Reaction of the pathway -

Reaction of HMP pathway is divided into Two phases .

1. Oxidative phase
2. Non-oxidative phase

1. Oxidative phase :-

Step 1 :- Glucose-6-phosphate Convert into 6-phosphogluconolactone in presence of enzyme Glucose-6-phosphate dehydrogenase . (Rate limiting step)

Step 2 : 6-phosphogluconolactone Convert into 6-phosphogluconate in presence of enzyme Gluconolactone hydrolase.

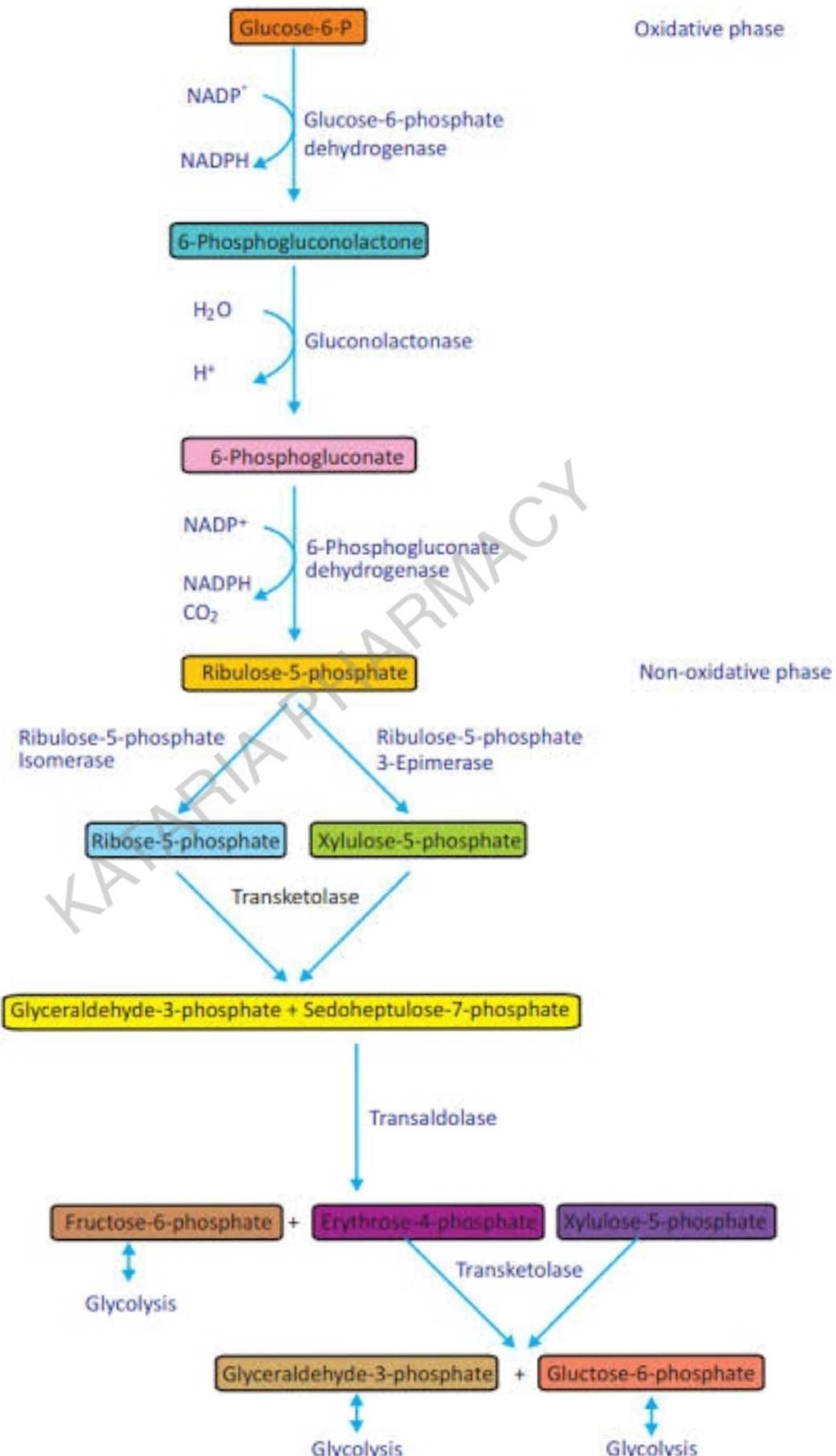
Step 3 : NADPH is again generated.

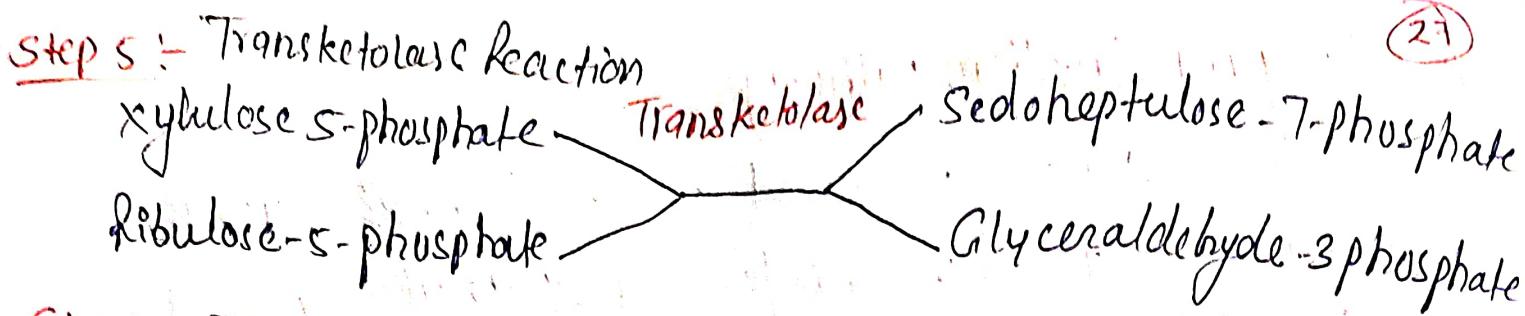
6-phosphogluconate Convert into Ribulose-5-phosphate in presence of enzyme 6-phosphogluconate dehydrogenase and 6-molecule of CO_2 is released.

② Non-oxidative phase:-

Step 4 :- Termination

Ribulose-5-phosphate Convert in
 Epimerase \rightarrow Xylulose-5-phosphate
 Isomerase \rightarrow Ribulose-5-phosphate





Step 6: Transaldolase Reaction -

Sedoheptulose-7-phosphate and Glyceraldehyde-3-phosphate
Convert into Erythrose-4-phosphate and Fructose-6-phosphate
in presence of enzyme Transaldolase.

#. Significance :-

- Free radical Scavenging
 - The free radicals (super oxide, hydrogen peroxide) are continuously produced in all cells.
- Hmp Shunt is unique in generating two important products - pentose and NADPH
- NADPH is required for the bio synthesis of fatty acids and steroids.
- NADPH is required by the RBC to keep the glutathione in the reduced state.

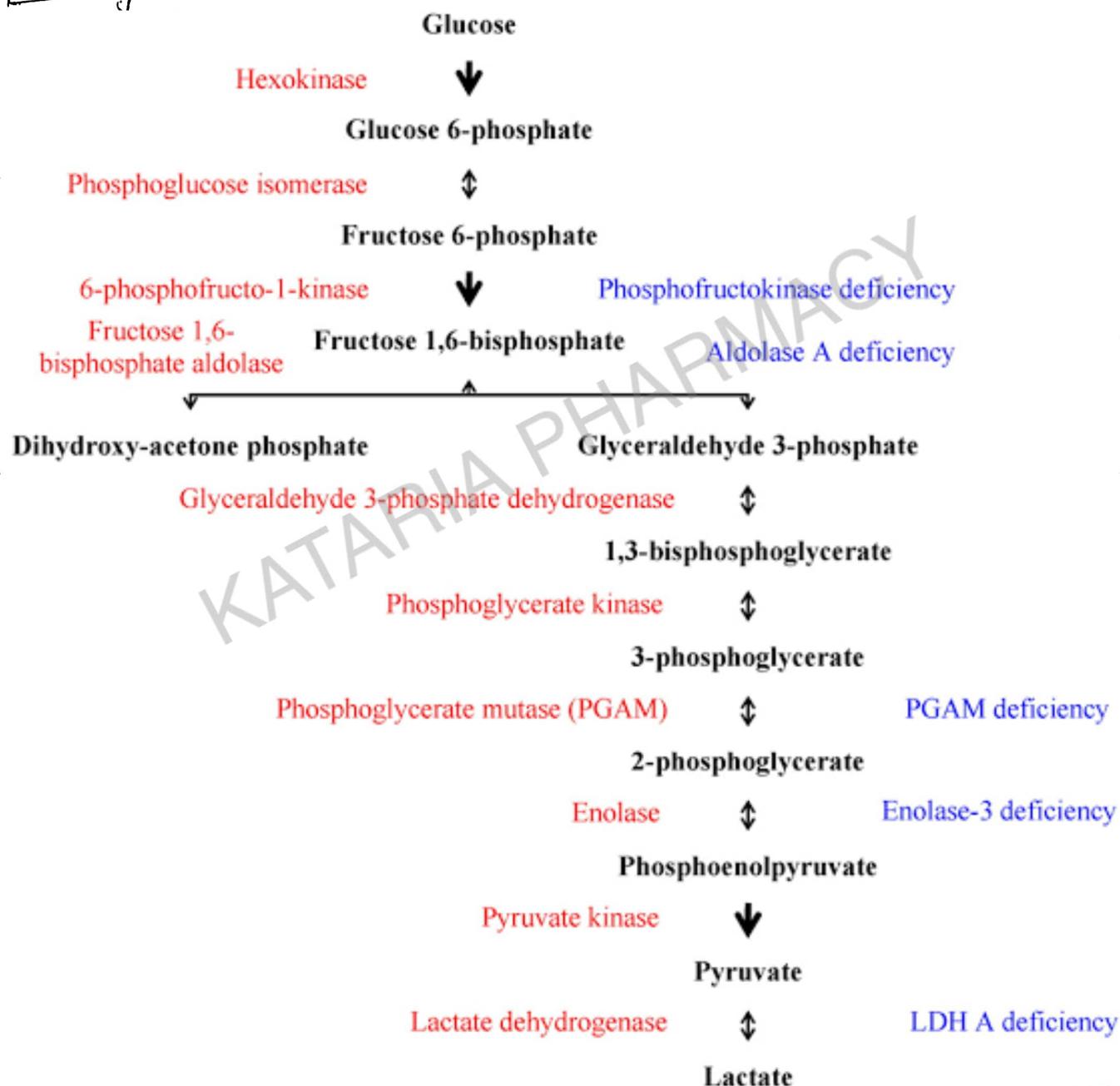
* Glycogen Metabolism pathways :-

- In animals, glucose is stored in the form of glycogen.
- Glycogen is stored in the form of granules in cell Cytosol where most of the enzyme required for the Synthesis and breakdown are found.

Glycogenesis :-

Glycogenesis is glycogen Synthesis from glucose. This occurs in Cytosol with the help of Adenosine Triphosphate and Triphosph Uridine triphosphate.

Pathway



* Glycogen Storage Disease (GSD) :-

- Glycogen storage disease are genetic disorder of glycogen metabolism occurring due to accumulation of large amount of glycogen or its metabolites into the tissue.
- Glycogen accumulates in the absence of glycogen metabolising enzyme. Not all but few of the disorders are serious.
- Glycogen storage diseases are as follows.

① Von Gierke's Disease :-

This disease is related to the deficiency or complete absence of enzyme glucose-6-phosphate in liver, kidney and intestine.

② Pompe's Disease :-

It is a fatal disease arising due to decrease in lysosomal α -glucosidase, thus glycogen is not utilised by lysosomes and accumulates in body tissue.

③ Anderson's disease :-

In this disease, an intermediate product of glycogenesis amylopectin deposits in the liver, spleen and heart.

④ Other :-

- ① Cori's disease
- ② McArdle's Syndrome
- ③ Hers Disease

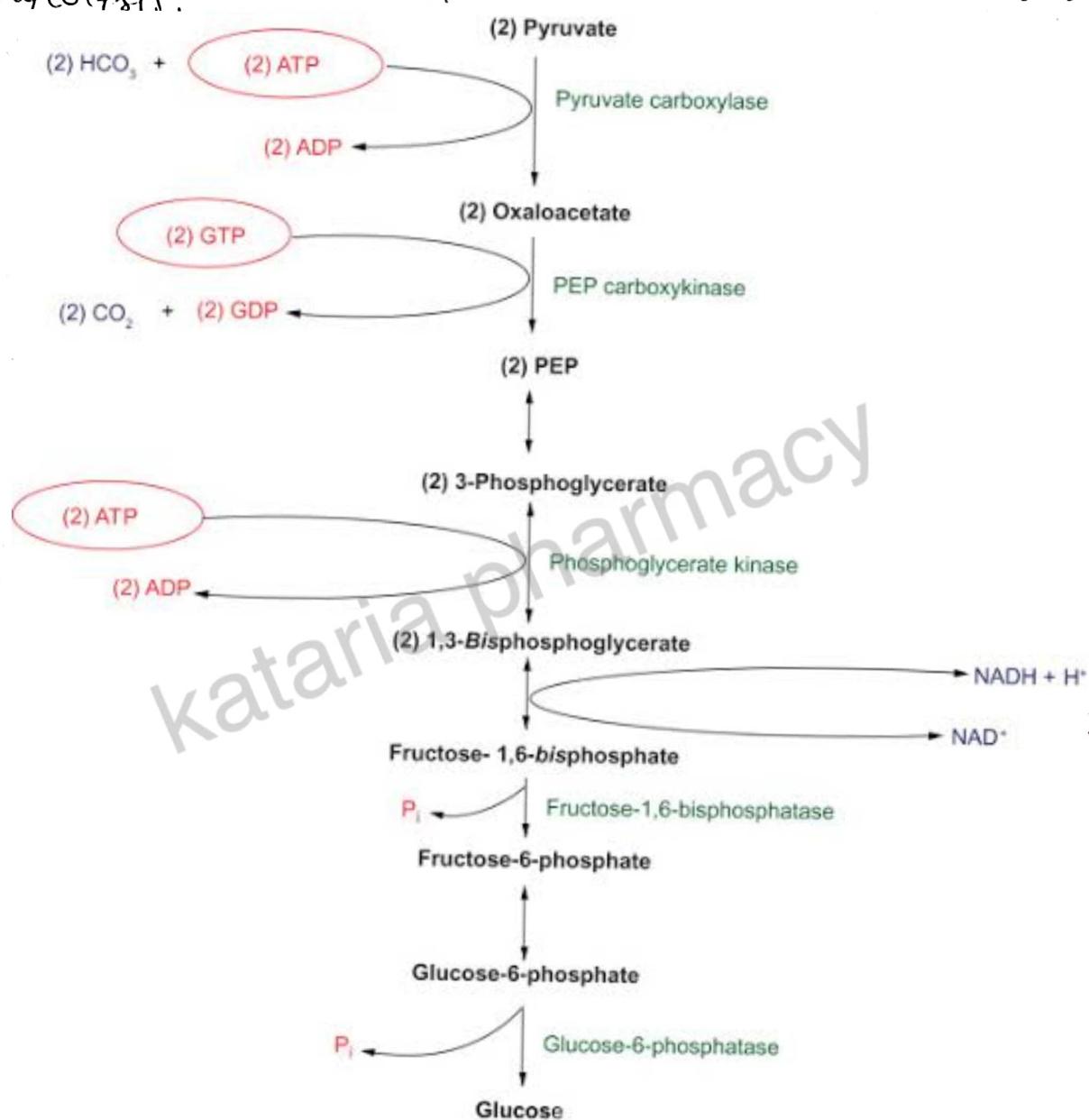
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A Gluconeogenesis

- Gluconeogenesis is a metabolic pathway which includes the biochemistry reaction Synthesising glucose from any non-Carbohydrate source such glycerol, pyruvate, lactate, and glucogenic amino acids.
- Most of the reactions of gluconeogenesis take place in the liver and few of the reaction in the Cortex part of kidney.

Pathway :-

Gluconeogenesis is almost the reversible pathway of glycolysis (Pyruvate converted to glucose) but not the exact reversal of glycolysis.



Significance of gluconeogenesis :-

- Gluconeogenesis meets the needs of the body for glucose & maintains glucose homeostasis in absence of carbohydrate.
- Some tissue such as the brain, RBCs, lens, cornea & kidney medulla requires continuous supply of energy.
- It is used to clear the products of the metabolism of other tissue from the blood e.g. lactate, glycerol.
- To maintain amount of glucose.
- Removal of glycerol produced by lipolysis.

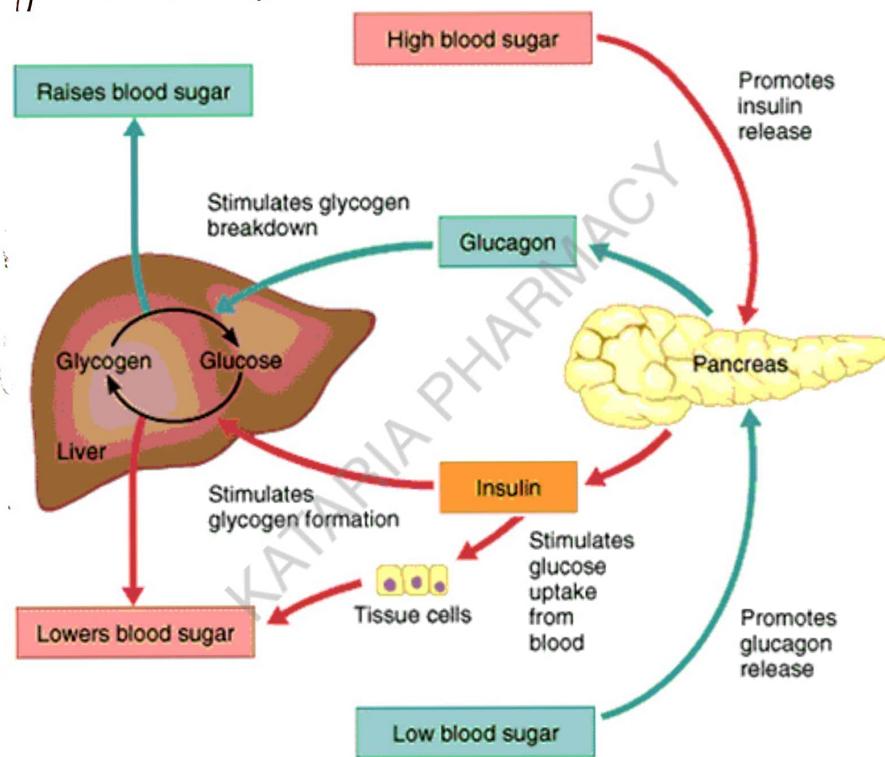
* Hormonal Regulation of Blood Glucose Level :-

There are two categories of endocrine influences.

1. Hormone which will decrease the blood glucose levels.

Eg. Insulin

2. Hormones which will increase the blood glucose levels.
e.g. Glucagon, Epinephrine, Cortisol and Glucocorticoids.

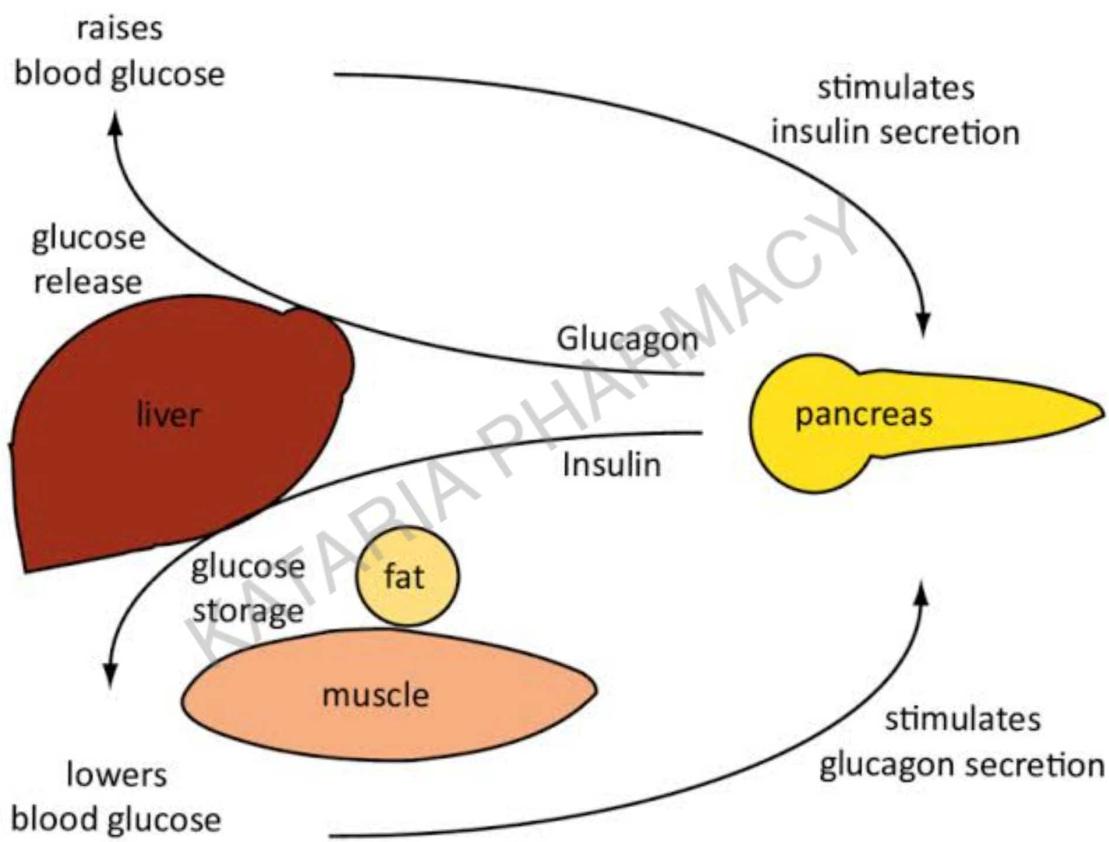


77 Insulin & Glucagon:

Both of these hormones are secreted by endocrine tissue of pancreas known as Islet Cells or Islets of Langerhans.

- Each Islet Contains two kinds of Cells:

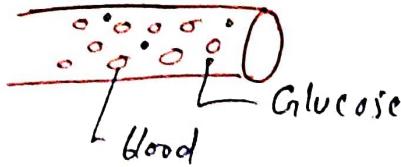
1. α -Cells (secrete glucagon) $\rightarrow \uparrow$ Blood glucose level.
2. β -Cells (secrete insulin) $\rightarrow \downarrow$ Blood glucose level



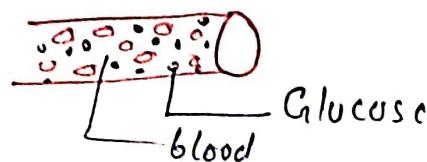
* Diabetes Mellitus:

Diabetes mellitus is a metabolic disorder in which a person has high blood sugar, either because the body does not produce enough insulin.

Normal Blood glucose level



High blood glucose



Types of DM :

① Type I DM :

This type is characterised by destruction of β -cell (insulin producing cell) and thus insulin is required for survival.

- Effects children

- Genetic variations & auto-immune response are leading cause.

② Type II DM :

This type is characterised by disorder of insulin action or insulin secretion. patient have relative insulin deficiency.

- It is also known as adult onset diabetes.

Causes of DM :

① Genetic factors → Genetic defect in β -cell function
→ Genetic defects in insulin action.

② Environmental factor → Obesity associated with modern living.
→ Lifestyle change (consumption of alcohol)

Symptoms of DM :

- ① Increase thirst
- ② Slow healing Cuts and Sores
- ③ Fatigue
- ④ Blurred vision.
- ⑤ Frequent urination.
- ⑥ Unexplained weight loss.

* Biological oxidation:

(34)

- In chemical terms, removal of electrons is termed as oxidation and addition of electrons is termed as reduction.
- Thus, oxidation always goes with reduction of an electron acceptor.
- This phenomenon of oxidation-reduction also applies in biochemical System and describe biological oxidation in a living System.
- In cell during respiration, biological oxidation takes place in four stages, Carbohydrate (glucose) is used as a source of energy.

1. Glycolysis: It is the first stage in which glucose degrades into pyruvate.

2. Oxidative Decarboxylation of pyruvate:

This stage occurs for a short term in which a 2-C containing molecule is formed from the 3-C containing molecule, pyruvate with the release of CO_2 .

3. Citric acid Cycle:

In this stage, the 2-C containing molecule is degraded into CO_2 .

* Electron transport Chain (ETC) :-

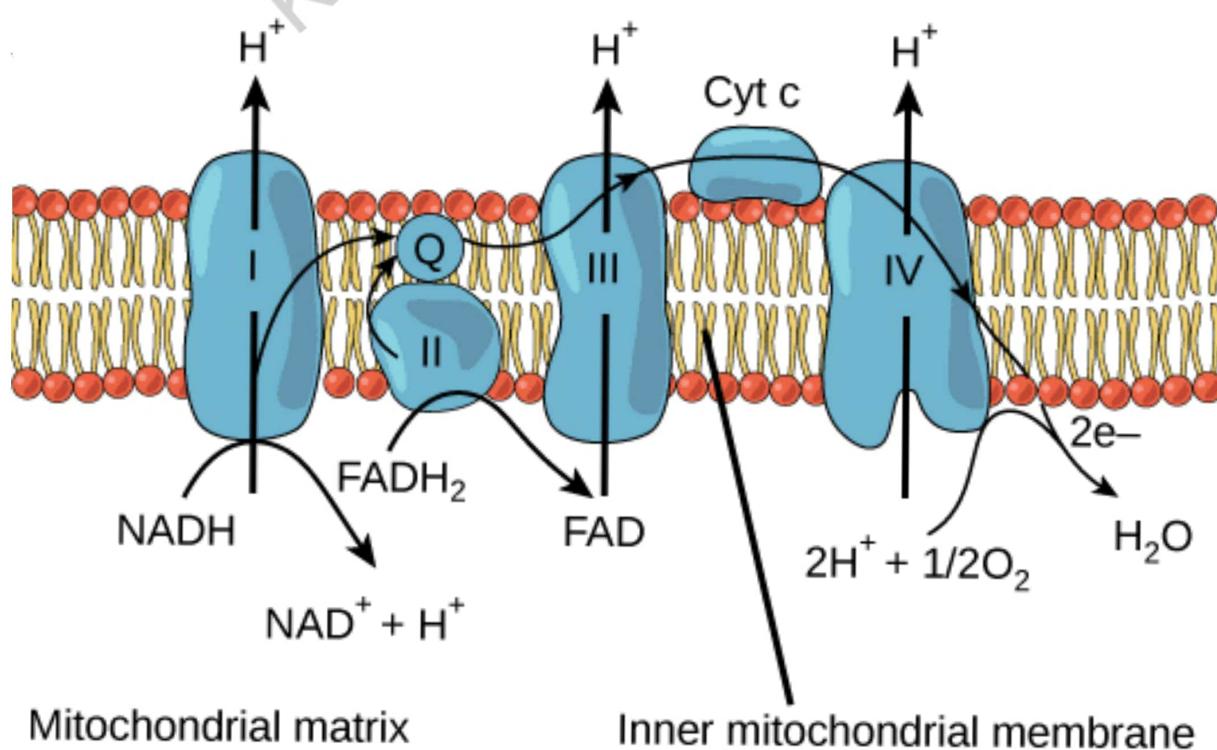
ETC Couple a chemical reaction b/w an electron donor and electron acceptor to the transfer of H^+ ions across a membrane, through a set of mediating biochemical reaction.

- These H^+ ions are used to produce ATP.
- In ETC, the electron travel through the chain, passing from a higher to a low energy level by moving through electron-rich to electron-deficit molecule.

ETC Components :-

- 5 protein Complexes - I, II, III, IV & V
- Coenzyme Q
- Cytochrome c - a peripheral membrane protein.

Intermembrane space



Mechanism of ETC :-

(B6)

The Steps involved in electron transport Chain are-

- ① The hydrogen Carriers from the citric acid Cycle enter the inner Mitochondrial Membrane folds known as Cristae.
- ② In the Cristae, the H⁺ ions are oxidised by a Series of step using molecular oxygen to release energy.
- ③ A part of this energy is utilised in oxidative phosphorylation in which ATP (from ADP) and inorganic phosphate are produced.
- ④ H⁺ ions or electron moves downhill i.e from a higher energy level to lower energy levels by passing from one carrier to the next, until they reach the oxygen.
- ⑤ The oxygen accepts electrons and reduces to form water.
- ⑥ At each transfer level, Some amount of energy is released in the form of heat which is utilised for producing ATP.

Significance :-

- ① Regenerating Electron Carrier
- ② Generating proton Gradient.

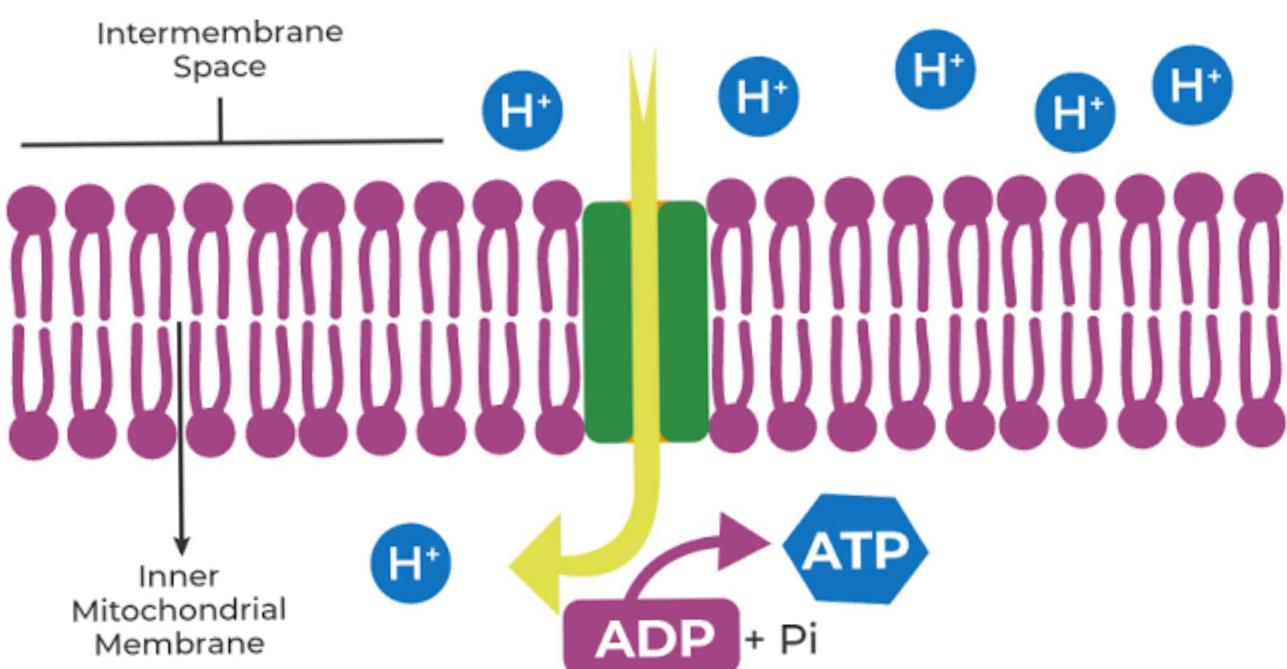
* Oxidative phosphorylation:

- ~~Oxidative phosphorylation is electron transport chain or chemiosmosis.~~
- It is a process in which ATP is formed as a result of transfer of electrons from NADH or FADH₂ to O₂ by series of electron carriers.

A. Oxidation Step : Electron transport Chain



B. Phosphorylation Step:



* Inhibitors of oxidative phosphorylation / uncouplers

- Electron transport within mitochondria and oxidative phosphorylation (ATP Synthesis) are combined process.
- It is also be said that oxidation and phosphorylation process proceed simultaneously.
- Some Compound have ability to uncouple these two reactions. and Such Compounds are known as Uncouplers.
- The Uncouplers increase permeability of the inner mitochondrial membrane towards protons (H^+) and thus inhibiting ATP formation.
- The energy generated during electron transport is dissipated in the form of heat.
- Thus, the Uncouplers facilitate oxidation of Substrates without ATP generation.

THANK YOU