JEE MAINS 2024



Exam Solutions Shift-02

Session-01 31 JANUARY 2024

IIT JEE

Time-03 Hrs

M.Marks : 300

Topic Covered Physics : Full Syllabus Chemistry : Full Syllabus Maths : Full Syllabus

GENERAL INSTRUCTION

1. Immediately fill in the particulars on this page of the test booklet.

2. The test is of 3 hours duration.

3. The test booklet consists of 90 questions. The maximum marks are 300.

4. There are Three Sections in the question paper, Section I, II & III consisting of Section-I (Physics), Section-II (Chemistry), Section-III (Mathematics) and having 30 questions in each part in which first 20 questions are compulsory and are of Objective Type and Last 10 questions are integers type in which you have to attempt 5 questions only.

5. There is only one correct response for each question.

6. Each correct answer will give 4 marks while 1 Mark will be deducted. 7. No student is allowed to carry any textual material, printed or written, bits of papers, pager, mobile phone, any electronic device, etc. inside the examination room/hall.

8. On completion of the test, the candidate must hand over the Answer Sheet to the Invigilator on duty in the Room/Hall. However, the candidates are allowed to take away this Test Booklet with them.

(Memory Based Solutions)

Physics

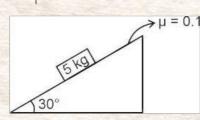
SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer:

1. For the block shown, F_1 is the minimum force required to move block upwards and F_2 is the minimum force required to prevent it from slipping,





- (1) 50 3 N
- (2) 5 🗸 N
- (3) 25 3 N

(4)
$$\frac{5\sqrt{3}}{2}$$
 N

Answer (2)

Sol. $f_{\kappa} = \mu mg \cos \theta$

$$= 0.1 \times \frac{50 \times \sqrt{3}}{2}$$
$$= 2.5 \sqrt{3} \text{ N}$$
$$F_1 = mg \sin\theta + f_K$$
$$= 25 + 2.5 \sqrt{3}$$
$$F_2 = mg \sin\theta - f_K$$
$$= 25 - 2.5 \sqrt{3}$$

- $\therefore F_1 F_2 = 5 \sqrt[8]{3} N$
- 2. Force on a particle moving in straight line is given by $\vec{F} = 6t^2\hat{i} - 3t\hat{j}$ and velocity is $\vec{v} = 3t^2\hat{i} + 6t\hat{j}$. Find power at t = 2.
 - (1) 216 W
 - (2) 108 W
 - (3) 0 W
 - (4) 54 W

Answer (1)



Sol. $P = \vec{F} \cdot \vec{v}$

 $=18t^4 - 18t^2$

- $\Rightarrow P(t=2)=18[16-4]=216 \text{ W}$
- 3. If $E = \frac{A x^2}{Bt}$ where *E* is energy, *x* is displacement

and t is time. Find dimensions of AB

- (1) $[M^{-1}L^2T]$
- (2) $[ML^2T^{-1}]$
- (3) $[M^{-1}L^2T^{-2}]$
- (4) [ML²T⁻²]

Answer (1)

Sol. $[A] = L^2$

$$B = \frac{x^2}{tE} = \frac{L^2}{TML^2T^{-2}} = \frac{1}{MT^{-1}}$$

[B] = M⁻¹T
[AB] = [M⁻¹L²T]

- Unpolarised light incident on transparent glass at incident angle 60°. If reflected ray is completely polarised, then angle of refraction is
 - (1) 45°
 - (2) 60°
 - (3) 30°
 - (4) 37°

Answer (3)

Sol. By Brewsters law

$$\therefore \quad 1 \times \frac{\sqrt{3}}{2} = \sqrt{3} \times \sin r$$

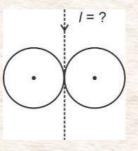
$$\sin r = \frac{1}{2}$$

2

 $r = 30^{\circ}$



 Two solid spheres each of mass 2 kg and radius 75 cm are arranged as shown. Find MOI of the system about the given axis.



- (1) 3.15 kg m²
- (2) 31.5 kg m²
- (3) 0.9 kg m²
- (4) 9 kg m²

Answer (1)

Sol. $I = \left(\frac{2}{5}MR^2 + MR^2\right) \times 2$

 $=\frac{14}{5}\times2\times\frac{9}{16}$

$$=\frac{63}{20}$$

= 3.15 kg m²

6. If the current through an incandescent lamp decreases by 20%, how much change will be there in its illumination?

(1) 36%	(2) 64%
(3) 20%	(4) 40%

Answer (1)

Sol. $p = i^2 R$

 $p' = 0.64 i^2 R$

7. Find the speed of sound in oxygen gas at STP.

(1) 300 m/s

- (2) 350 m/s
- (3) 330 m/s
- (4) 400 m/s

Answer (3)

Sol.
$$v = \sqrt{\frac{\gamma RT}{M}} = 330 \text{ m/s}$$

 Find average power in electric circuit if source voltage (V) = 20sin(100ωt) and current in the circuit

$$(l) = 2\sin(100\omega t + \frac{\pi}{3})$$
(1) 10 W (2) 20 W
(3) 5 W (4) 15.5 W

Answer (1)

Sol. <P> = IV coso

$$=\frac{20}{\sqrt{2}}\times\frac{2}{\sqrt{2}}\times\cos 60^{\circ}$$

= 10 W

- 9. In a photoelectric experiment, frequency $f = 1.5f_0$ (f_0 : threshold frequency). If the frequency of light is changed to f/2, then photocurrent becomes (intensity of light has doubled)
 - (1) Zero
 - (2) Doubled
 - (3) Same
 - (4) Thrice

Answer (1)

Sol. Since
$$\frac{f}{2} < f_0$$

 \Rightarrow current = 0

- Radius of curvature of equiconvex lens is 20 cm. Material of lens is having refractive index of 1.5. Find image distance from lens if an object is placed 10 cm away from the lens.
 - (1) 20 cm
 - (2) 10 cm
 - (3) 40 cm
 - (4) 5 cm

Answer (1)

Sol.
$$\frac{1}{f} = (\mu - 1) \left(\frac{2}{R}\right)$$
 $f = 20 \text{ cm}$
 $\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$
 $\frac{1}{v} + \frac{1}{u} = \frac{1}{v}$

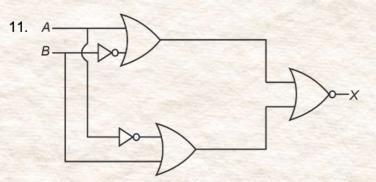
v 10 20







(Memory Based Solutions)



Draw truth table of given gate circuit.

	A	B	x	(2)	A	В	X
(1)	0	0	0	(2)	0	0	0
	0	B 0 1 0 1	0		0	1	0
	1	0	0		1	0	0
	1	1	1		1	1	0
			V				1.11
(0)	A	B	x	11	Α	В	X
(3)	A 0	B 0	x 1	(4)	A 0	B 0	<i>X</i> 1
(3)	A 0 0	B 0 1	<i>X</i> 1 0	(4)	A 0 0	B 0 1	<i>X</i> 1 0
(3)	A 0 0 1	B 0 1 0	X 1 0 0	(4)	A 0 0 1	B 0 1 0	X 1 0 0
(3)	A 0 0 1 1	<i>B</i> 0 1 0 1	X 1 0 0 0	(4)	A 0 0 1	B 0 1 0 1	X 1 0 1

Answer (2)

Sol. $X = (\overline{A + \overline{B}}) + (\overline{A} + B)$

 $\left(\overline{\overline{A} + \overline{B}}\right) \left(\overline{\overline{\overline{A}} + B}\right)$ $\left(\overline{\overline{A} \cdot \overline{B}}\right) \left(\overline{\overline{\overline{A}} \cdot \overline{B}}\right)$ $\left(\overline{\overline{A} \cdot B}\right) \cdot \left(\overline{A \cdot \overline{B}}\right) = \overline{\overline{A} \cdot B} \cdot \overline{A \cdot \overline{B}} = 0$

12. The magnetic flux through a loop varies with time as $\phi = 5t^2 - 3t + 5$. If the resistance of loop is 8 Ω , find the current through it at *t* = 2 s

(1)
$$\frac{15}{8}$$
 A (2) $\frac{5}{8}$ A
(3) $\frac{17}{8}$ A (4) $\frac{13}{8}$ A

Answer (3)

Sol. $\frac{d\phi}{dt} = 10t - 3$ at t = 2, V = 17 $i = \frac{V}{R} = \frac{17}{8}$ A

- 8 moles of oxygen and 4 moles of nitrogen are at same temperature *T* and are mixed. The total internal energy is
 - (1) 60RT
 - (2) 15RT
 - (3) 30RT
 - (4) 90RT

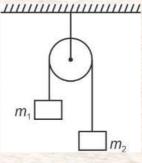
Answer (3)

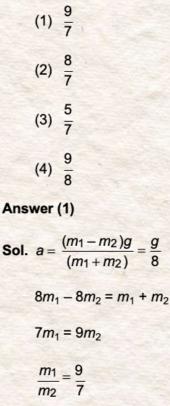
Sol.
$$U = nC_v T$$

$$\Rightarrow U = n_1 C_V T + n_2 C_V T$$

$$\Rightarrow 8 \times \frac{5R}{2} \times T + 4 \times \frac{5R}{2} \times T$$
$$= 30RT$$

14. In the system shown below, the pulley 4 string are ideal. If the acceleration of blocks is $\frac{g}{8}$, find $\frac{m_1}{m_2}$







15. The force between two charged particle placed in air at separation x is F_0 . Both the charged particle immerged in a medium of dielectric constant K without changing separation between two charge, then net force on one of the particle is now

(1)
$$\frac{F_0}{K}$$

(2) $\frac{F_0}{2K}$ (3) $\frac{2F_0}{K}$

(4) Fo

Answer (1)

Sol. In air $F = \frac{1}{4\pi \epsilon_0} \frac{q_1 q_2}{r_2}$

In medium
$$F' = \frac{1}{4\pi(k \in 0)} \frac{q_1 q_2}{r^2}$$

$$F' = \frac{F_0}{K}$$

- 16. Two vector each of magnitude A are inclined at angle θ with each other, then magnitude of resultant vector is
 - (1) $A\cos^2\frac{\theta}{2}$ (2) $2A\cos\frac{\theta}{2}$ (3) 2A cosθ (4) $A\cos\frac{\theta}{2}$

Answer (2)

Sol. The magnitude of resultant vector $= \sqrt{a^2 + b^2 + 2ab\cos\theta}$ here a = b = Athen $R = \sqrt{A^2 + A^2 + 2A^2 \cos \theta}$ $= A\sqrt{2}\sqrt{1+\cos\theta}$ $= \sqrt{2}A \sqrt{2\cos^2\frac{\theta}{2}}$ = $2A\cos\frac{\theta}{2}$

17. Statement 1 : Electric and magnetic energy density in electromagnetic waves are equal.

Statement 2 : Electromagnetic waves exert pressure on a surface.

- (1) Statement 1 is true & Statement 2 is true and is correct explanation of Statement 1
- (2) Statement 1 is true & Statement 2 is true but is not correct explanation of Statement 1
- (3) Statement 1 is true but Statement 2 is false
- (4) Statement 1 is false but Statement 2 is true

Answer (2)

$$\textbf{Sol.} \quad \frac{1}{2}\varepsilon_0 E^2 = \frac{B^2}{2\mu_0}$$

$$\therefore E = CB \text{ and } C = \frac{1}{\mu_0 \varepsilon_0}$$

- 18. A pendulum completes 50 oscillations in 40 seconds. If the length of pendulum is (20 ± 0.2) cm and resolution of watch is 1 second, find the percentage error in calculation of g.
 - (1) 7%
 - (2) 3%
 - (3) 6%
 - (4) 4%

Answer (3)

(R)

19

20

Sol.
$$T = 2\pi \sqrt{\frac{T}{g}}$$

 $g = \frac{4\pi^2 i}{T^2}$
 $\frac{\Delta g}{g} = \frac{\Delta i}{l} + \frac{2\Delta T}{T}$
 $= \frac{0.2}{20} + 2\left(\frac{1}{40}\right)$
 $= 6\%$
19.
20.

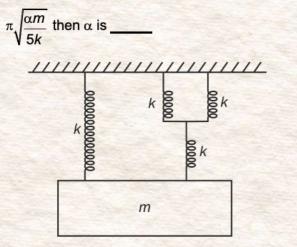


SCOLASS BOOST

SECTION - B

Numerical Value Type Questions: This section contains 10 Numerical based questions. The answer to each question should be rounded-off to the nearest integer.

21. The period of oscillation of system shown below is



Answer (12)

Sol. $k_{eq} = \frac{2k \cdot k}{3k} + k = \frac{5k}{3}$

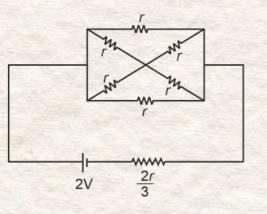
Angular frequency of oscillation (ω) = $\sqrt{\frac{k_{eq}}{m}}$

$$\omega = \sqrt{\frac{5k}{3m}}$$

Period of oscillation (τ) = $\frac{2\pi}{\omega} = 2\pi \sqrt{\frac{3m}{5k}}$

$$=\pi\sqrt{\frac{12m}{5k}}$$

22. In the given circuit, $r = 2 \Omega$. The power dissipated in the circuit is _____W.



Sol. $R_{eq} = r$

$$P = \frac{V^2}{r} = \frac{4}{2} = 2 \text{ W}$$

23. A body of mass *m* is projected with speed *u* at angle 45° with horizontal. The angular momentum of body, about point of projection when body is at

highest point, is
$$\frac{\sqrt{2} m u^3}{xg}$$
 find x,

Answer (8)

Sol.
$$L = mu \cos \theta \frac{u^2 \sin^2 \theta}{2g}$$

$$= mu^3 \frac{1}{4\sqrt{2}g} \Rightarrow x = 8$$

24. Mass of moon is $\frac{1}{81}$ times the mass of a planet and

radius is $\frac{1}{9}$ times the radius of the planet. The ratio of escape speed from planet to escape speed from moon is _____.

Answer (3)

Sol.
$$v_{\rm esc} = \sqrt{\frac{2 \, GM}{R}}$$

$$\Rightarrow$$
 Ratio = $\sqrt{\frac{81}{9}} = 3$

25. Find the mass number of an atom whose radius is half of that of a given atom of mass number 192.

Sol.
$$r = R_0 (192)_3^{\frac{1}{3}}$$

 $\frac{r}{2} = R_0 (m)^{\frac{1}{3}}$
 $m = \frac{192}{8} = 24$

26.

Pending

30.

Answer (2)



(Memory Based Solutions)



Chemistry Statement-I: Among 15th group hydrides reducing 3. character decreases from NH₃ to BiH₃. Statement-II: E2O3 and E2O5 are always basic. [Where E is group 15 element] (1) Both statement-I and Statement-II are correct (2) Statement-I is correct and Statement-II is false (3) Statement-I is false and Statement-II is correct (4) Both Statement-I and Statement-II are false Answer (4) Sol. Reducing character increases from NH₃ to BiH₃. Group 15 oxides of type E2O3 and E2O5 are not always basic. 4. Which of the following has maximum ionic character? (2) AgCI (1) KCI (4) BaCl₂ (3) CoCl₂ Answer (1) Ovidation Charge Sol. Polarisation power ∝ Size for K⁺, polarising power is least and ionic character is maximum. 5. Match the following : (a) [Cr(H₂O)₆]⁺³ (i) t²_{2a}eg° (ii) t³₂₀eg° (b) [Fe(H₂O)₆]⁺³ (iii) t³₂₀eg² (c) [Ni(H₂O)₆]⁺² (d) [V(H₂O)₆]⁺³ (iv) $t_{2a}^6 eg^2$ (1) (a)-(ii), (b)-(iii), (c)-(iv), (d)-(i) (2) (a)-(iii), (b)-(iv), (c)-(i), (d)-(ii) (3) (a)-(iv), (b)-(ii), (c)-(iii), (d)-(i) (4) (a)-(ii), (b)-(iv), (c)-(i), (d)-(iii)

Answer(1)

Sol. (a) $[Cr(H_2O)_6]^{+3} \rightarrow Cr^{+3} \rightarrow t_{2d}^3 eg^\circ$

(b) $[Fe(H_2O)_6]^{+3} \rightarrow Fe^{3+} \rightarrow t^3_{26}eg^2$

(c) $[Ni(H_2O)_6]^{+2} \rightarrow Ni^{2+} \rightarrow t_{2g}^6 eg^2$

(d) $[V(H_2O)_6]^{+3} \rightarrow V^{3+} \rightarrow t^2_{2d}eg^\circ$

SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which ONLY ONE is correct.

Choose the correct answer :

1. Statement 1: S8 disproportionate into H2S2O3 and S²⁻ in alkaline medium

Statement 2 : CIO₄ undergoes disproportionation in acidic medium.

- (1) Statement 1 is correct but statement 2 is incorrect
- (2) Statement 1 is incorrect but statement 2 is correct
- (3) Both statement 1 and statement 2 are correct
- (4) Both statement 1 and statement 2 are incorrect

Answer (1)

Sol. (1)
$$S_{8}^{0}$$
 + NaOH \rightarrow Na₂⁻² + Na₂S₂O₃
Reduction

(2) Cl is in its highest oxidation state (+7). It cannot be further oxidised

Therefore, statement 1 is correct but statement 2 is incorrect.

- 2. Which of the following is correct?
 - (1)[NiCl4]²⁻ diamagnetic

[Ni(CO)₄] – diamagnetic

- (2)[Ni(CO)4] diamagnetic
- [NiCl4]2- paramagnetic
- (3)[NiCl₄]²⁻ paramagnetic
 - [Ni(CO)₄] paramagnetic
- (4)[NiCl4]²⁻ paramagnetic
 - [Ni(CO)₄] diamagnetic

Answer (2)

Sol. Ni²⁺ : 4s⁰3d⁸ (No pairing with Cl⁻) [Ni(CO)₄]: 4s⁰3d¹⁰ (diamagnetic)

 Quantum number for outermost electron of K-atom are given by

(1)
$$n = 4, l = 0, m = 0, s = \frac{1}{2}$$

(2) $n = 4, l = 1, m = 0, s = \frac{1}{2}$
(3) $n = 3, l = 0, m = 0, s = \frac{1}{2}$
(4) $n = 4, l = 0, m = 1, s = \frac{1}{2}$

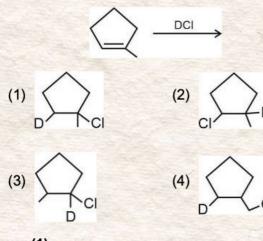
Answer (1)

Sol. K₁₉ = 1s²2s²2p⁶3s²3p⁶4s¹

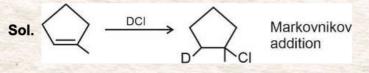
For 4s electron

n = 4 I = 0

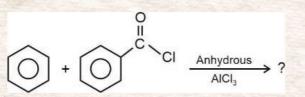
- m = 0
- $s=\frac{1}{2}$
- 7. What is the product formed in the below given reaction?



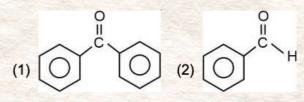
Answer (1)



8. What is the major product formed in the following reaction?

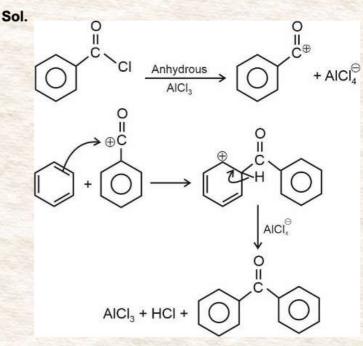




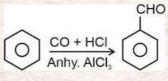




Answer (1)



9. Identify the given rection



- (1) Rosenmund reaction
- (2) Stephen reaction
- (3) Gattemann Koch reaction
- (4) Etard reaction

Answer (3)

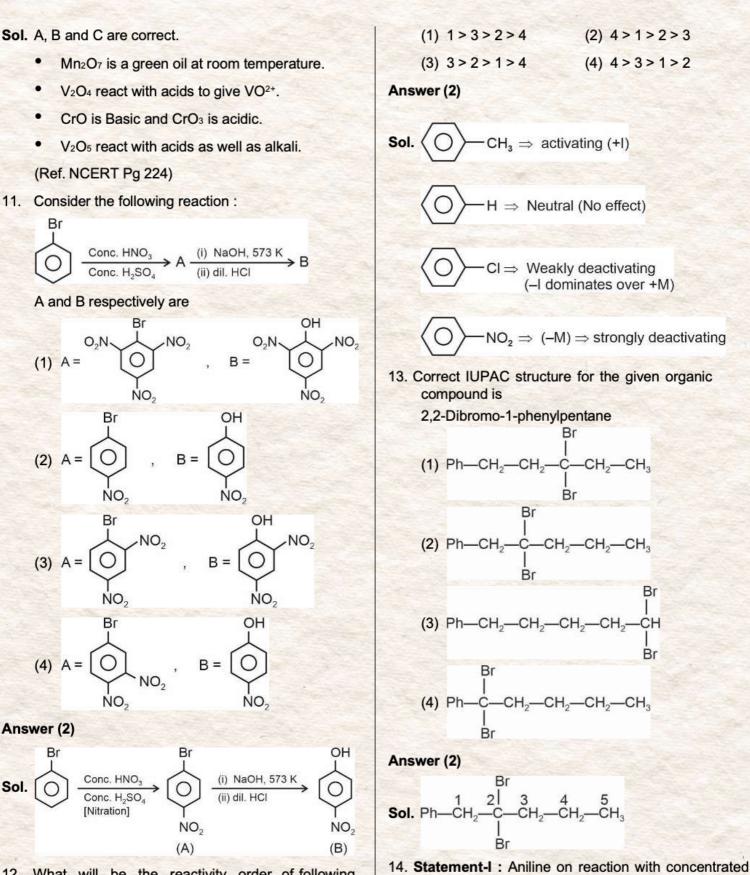
- Sol. The given reaction is Gattemann Koch reaction.
- 10. Choose the correct answers.

(A)Mn₂O₇ is a oil at room temperature.

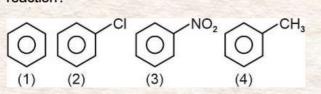
- (B)V₂O₄ react with acid to give VO²⁺
- (C) CrO is a basic oxide
- (D)V₂O₅ does not react with acids.
- (1) A, B and C only
- (2) B, C and D only
- (3) A only
- (4) B and C only

Answer (1)

STOLAS BOOST



12. What will be the reactivity order of following compounds towards electrophilic substitution reaction?



Statement-II : Aniline forms a salt with anhydrus AICI₃ in Friedel Craft's reaction.

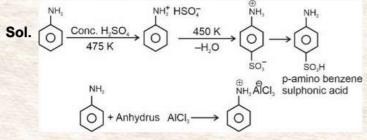
H₂SO₄ at 475 K gives p-amino benzene sulphonic

acid. This gives blood red colour with Lassaigne's

test.

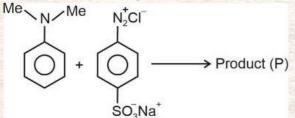
- (1) Both Statement-I and Statement-II are correct
- (2) Both Statement-I and Statement-II are incorrect
- (3) Statement-I is correct and Statement-II incorrect
- (4) Statement-I is incorrect and Statement-II correct

Answer (1)

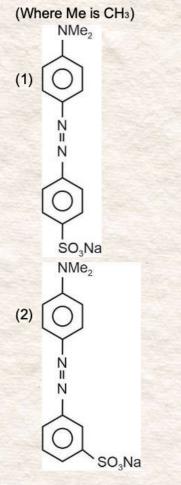


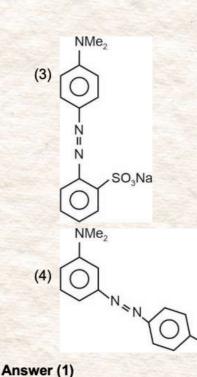
p-amino benzene sulphonic acid contains both N and S, so it gives blood red colour with Lassaigne's test.

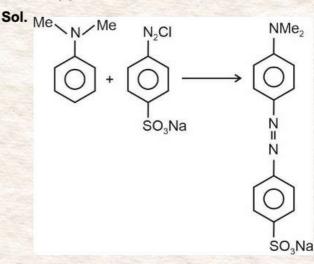
15. Consider the following reaction.



Select P







is an example of azo coupling reaction and final product is methyl orange.

SO₃Na

16. A(g)
$$\longrightarrow$$
 B(g) + $\frac{1}{2}$ C(g)

In the about reaction, the correct relation between $K_{\text{p}},\,\alpha$ and equilibrium pressure (p) is

(1)
$$K_{p} = \frac{\alpha^{\frac{1}{2}} 2p^{\frac{1}{2}}}{(2+\alpha)^{\frac{1}{2}}}$$

(2) $K_{p} = \frac{\alpha^{\frac{1}{2}} p^{\frac{3}{2}}}{(2+\alpha)^{\frac{3}{2}}}$
(3) $K_{p} = \frac{\alpha^{\frac{1}{2}} 2p^{\frac{1}{2}}}{(2+\alpha)^{\frac{3}{2}}}$
(4) $K_{p} = \frac{\alpha^{\frac{3}{2}} p^{\frac{1}{2}}}{(2+\alpha)^{\frac{1}{2}} (1-\alpha)}$

Answer (4)



Sol.

$$A(g) \iff B(g) + \frac{1}{2}(g)$$
Initial n 0 0
moles
Eqb. $n(1 - \alpha)$ $n\alpha$ $\frac{n\alpha}{2}$
total moles = $n(1+\alpha)_{2}$
Eqb. $\frac{(1 - \alpha)p}{1 + \alpha}$ $\frac{\alpha p}{1 + \alpha}$ $\frac{\left(\frac{\alpha}{2}\right)p}{1 + \alpha}$

$$K_{p} = \frac{\alpha p}{\left(1 + \frac{\alpha}{2}\right)} \times \left[\frac{\alpha p}{\left[\left(2 + \alpha\right)^{2}\right]^{2}}\right]$$

$$\frac{(1 - \alpha)p}{1 + \frac{\alpha}{2}}$$

$$K_{p} = \frac{\alpha^{3/2} p^{\frac{1}{2}}}{(2 + \alpha)^{\frac{1}{2}}(1 - \alpha)}$$
Pending

SECTION - B

Numerical Value Type Questions: This section contains 10 Numerical based questions. The answer to each question should be rounded-off to the nearest integer.

21. Half life of a first order reaction is 36 hr. Find out time (in hr) required for concentration of reactant to get reduced by 90%.

Answer (120)

17.

18.

19.

20.

Sol.
$$t_{90} = \frac{2.303}{k} \log \left(\frac{100}{100 - 90} \right)$$

= $\frac{2.303 \times 36}{2.303 \times \log 2} \times \log 10 = \frac{36}{0.3} = 120$

22. A 1 mol ideal gas expands from 10 L to 100 L at 300 k, if above expansion takes place reversibly and isothermally then magnitude of work done is (in KJ)

Answer (06)

Sol. w = -nRT ln $\frac{V_2}{V_1}$ $|w| = 2.303 \text{ nRT } \log \frac{V_2}{V_1}$



$$|\mathbf{w}| = 1 \times 2.303 \times 8.314 \times 300 \log \frac{100}{10}$$

 $|w| = 5.744 \text{ kJ} \approx 6 \text{ kJ}$

23. How many of the following vitamins are stored in Human Body?

A, B, C, D, E, K?

Answer (4)

Sol. A, D, E, K vitamins are fat soluble vitamins, are stored in liver and adipose tissue.

While vitamin B and vitamin C are water soluble and must be supplied regularly in diet (not stored) (NCERT, Pg: 426) (except vitamin B12)

24. Number of moles of H⁺ required by 1 mole MnO₄⁻ to oxidize oxalate ion to CO2 is

Answer (8)

Sol. The balanced reaction is as follows $2MnO_{4}^{-} + 5C_{2}O_{4}^{2-} + 16H^{+} \rightarrow 2Mn^{2+} + 10CO_{2} + 8H_{2}O_{4}$

2 mole MnO₄ react with 16 mole H⁺

1 mole MnO₄ will react with 8 mole H⁺

25. The potassium chloride is heated with potassium dichromate and conc. sulphuric acid to give products. The oxidation state of chromium in product is (+)_

Answer (06.00)

Sol. This is an example of chromyl chloride test

$$K_2Cr_2O_7 + 4KCl + 6H_2SO_4 \rightarrow 6KHSO_4$$

+ 2CrO2Cl2 + 3H2O

Oxidation state of Cr is +6.

26. Number of structural isomeric products formed by monochlorination of 2-methylbutane in presence of sunlight is

Answer (4)

Sol
$$H_{3}C - CH_{2} - CH - CH_{3} \xrightarrow{\text{Monochlorination}} H_{3}C - CH_{2} - \overset{\bullet}{C}H - CH_{2}CI \xrightarrow{I}_{I} H_{3}C - CH_{2} - \overset{\bullet}{C}H - CH_{2}CI \xrightarrow{I}_{I} H_{3}C - CH_{2} - CCI - CH_{3} \xrightarrow{I}_{I} H_{3}C - CH_{2} - CCI - CH_{3} \xrightarrow{I}_{I} H_{3}C - \overset{\bullet}{C}HCI - CH - CH_{3} \xrightarrow{I}_{CH_{3}} H_{3}C - \overset{\bullet}{C}HCI - CH - CH_{3} \xrightarrow{I}_{CH_{3}} CH_{3} CIH_{2}C - CH_{2} - CH - CH_{3} \xrightarrow{I}_{CH_{3}} H_{3}C - \overset{\bullet}{C}H_{2} - CH_{2} - CH_{2} - CH_{3} \xrightarrow{I}_{CH_{3}} H_{3}C - \overset{\bullet}{C}H_{3} - CH_{3} \xrightarrow{I}_{CH_{3}} H_{3}C - \overset{\bullet}{C}H_{3} - \overset{$$

Pending

30.

28.



(Memory Based Solutions)

Mathematics

SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer :

- 1. $a = \sin^{-1}(\sin 5), b = \cos^{-1}(\cos 5)$ then $a^2 + b^2$ is equal to
 - (1) $8\pi^2 40\pi + 50$ (2) $4\pi^2 + 25$ (3) $8\pi^2 - 50$ (4) $8\pi^2 + 40\pi + 50$

Answer (1)

Sol. $a = \sin^{-1}(\sin 5) = 5 - 2\pi$

and
$$b = \cos^{-1}(\cos 5) = 2\pi - 5$$

 $\therefore a^2 + b^2 = (5 - 2\pi)^2 + (2\pi - 5)^2$
 $= 8\pi^2 - 40\pi + 50$

2. A coin is biased such that head has two chances than tails, what is the probability of getting 2 heads and 1 tail?

(1)	1 29	(2)	2 29
(3)	<u>1</u> 9	(4)	4 9

Answer (4)

Sol. Let probability of tail is $\frac{1}{3}$

- \Rightarrow Probability of getting head = $\frac{2}{3}$
- ... Probability of getting 2 heads and 1 tail

$$= \left(\frac{2}{3} \times \frac{2}{3} \times \frac{1}{3}\right) \times 3$$
$$= \frac{4}{27} \times 3$$
$$= \frac{4}{9}$$

3. Let mean and variance of 6 observations a, b, 68, 44, 40, 60 be 55 and 194. If a > b then find a + 3b

(1) 211.83	(2) 201.59
(3) 189.57	(4) 198.87

Answer (2)

Sol. $\frac{a+b+68+44+40+60}{6} = 55$ 212+a+b=330 $\Rightarrow a+b=118$ $\frac{\sum x_i^2}{n} - (\overline{x})^2 = 194$

$$\frac{a^2 + b^2 + (68)^2 + (44)^2 + (40)^2 + (60)^2}{6} - (55)^2 = 194$$

= 3219
11760 +
$$a^2$$
 + b^2 = 19314
 $\Rightarrow a^2 + b^2$ = 19314 - 11760
= 7554
($a + b$)² -2 ab = 7554
From here b = 41.795
 $a + b$ = 118
 $\Rightarrow a + b + 2b$ = 118 + 83.59
= 201.59

4. If 2nd, 8th, 44th terms of A.P. are 1st, 2nd and 3rd terms respectively of G.P. and first term of A.P. is 1 then the sum of first 20 terms of A.P. is

(1) 970	(2) 916

(3) 980	(4) 990
1-7	(-)

Answer (1)

Sol. *a* + *d*, *a* + 7*d* and *a* + 43*d* are 1st, 2nd, 3rd term of G.P.

3

$$\frac{a+7d}{a+d} = \frac{a+43d}{a+7d}$$

$$\Rightarrow (a+7d)^2 = (a+d)(a+43d)$$

$$\Rightarrow a^{2}+49d \neq 14d = a \neq 44ad + 43d$$

$$\Rightarrow 6d^2 = 30ad$$

$$\Rightarrow d^2 = 5d$$

$$\Rightarrow d = 0, 5$$

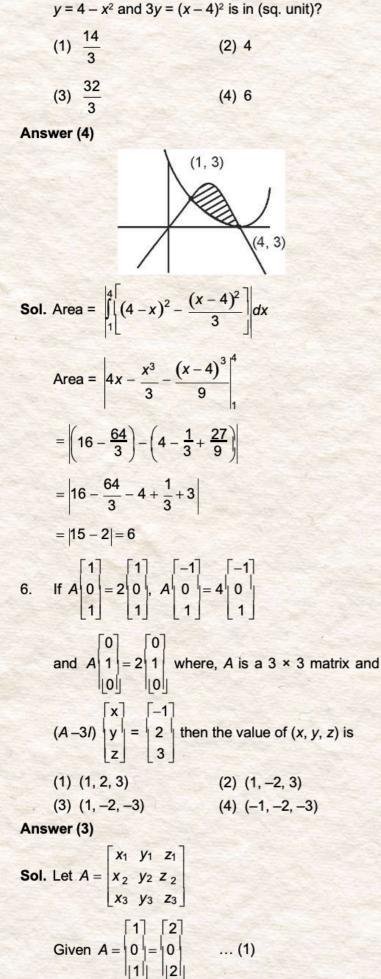
$$a = 1, d = 5$$

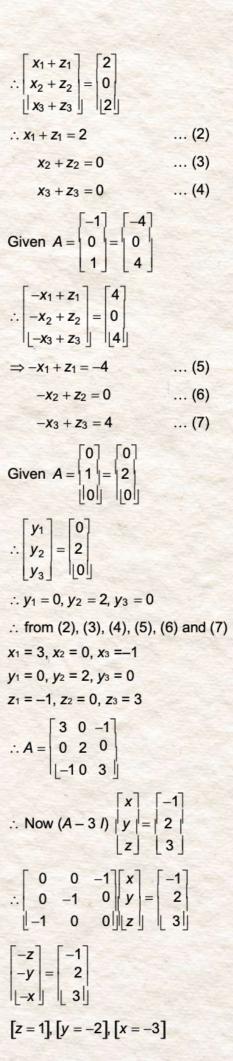
$$S_{20} = \frac{20}{2}[2+(19)5]$$

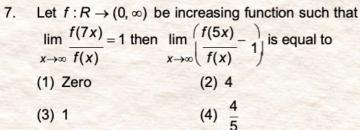
$$= 10[95+2]$$

$$= 970$$

The area of the region enclosed by the parabolas







(3) 1

Answer (1)

Sol. f is increasing function

x < 5x < 7x5x 7x f(x) < f(5x) < f(7x) $\frac{f(x)}{f(x)} < \frac{f(5x)}{f(x)} < \frac{f(7x)}{f(x)}$ $\lim_{x \to \infty} \frac{f(x)}{f(x)} < \lim_{x \to \infty} \frac{f(5x)}{f(x)} < \lim_{x \to \infty} \frac{f(7x)}{f(x)}$ $1 < \lim_{x \to \infty} \frac{f(5x)}{f(x)} < 1 \quad \Rightarrow \lim_{x \to \infty} \frac{f(5x)}{f(x)} = 1$ $\lim_{x\to\infty} \left(\frac{f(5x)}{f(x)} - 1\right) = 0$

Let z1 and z2 be two complex numbers such that 8. $z_1 + z_2 = 5$ and $z_1^3 + z_2^3 = 20 + 15i$, then the value of $|z_1^4 + z_2^4|$ is equal to

(1) 75	(2)	25√5
(3) 15√15	(4)	30√3

Answer (1) Sol. Z1 + Z2 = 5

$$\begin{aligned} z_1^3 + z_2^3 &= 20 + 15i \\ z_1^3 + z_2^3 &= (z_1 + z_2)^3 - 3z_1 z_2 (z_1 + z_2) \\ z_1^3 + z_2^3 &= 125 - 3z_1 \cdot z_2 (5) \\ \Rightarrow & 20 + 15i = 125 - 15z_1 z_2 \\ \Rightarrow & 3z_1 z_2 = 25 - 4 - 3i \\ & 3z_1 z_2 = 21 - 3i \\ & z_1 \cdot z_2 = 7 - i \\ & (z_1 + z_2)^2 &= 25 \\ & z_1^2 + z_2^2 &= 25 - 2(7 - i) \\ &= 11 + 2i \\ & \left(z_1^2 + z_2^2\right)^2 &= 121 - 4 + 44i \end{aligned}$$



$\Rightarrow z_1^4 + z_2^4 + 2(7-i)^2 = 117 + 44i$
$\Rightarrow z_1^4 + z_2^4 = 117 + 44i - 2(49 - 1 - 14i)$
= 21 + 72i
\Rightarrow $ z_1^4 + z_2^4 = 75$
9. The number of solutions of equation $e^{\sin x} - 2e^{-\sin x} = 2$ is
(1) More than 2 (2) 2
(3) 1 (4) 0
Answer (4)
Sol. Take $e^{\sin x} = t (t > 0)$
$\Rightarrow t - \frac{2}{t} = 2$
$\Rightarrow \frac{t^2 - 2}{t} = 2$
$\Rightarrow t^2 - 2t - 2 = 0$
$\Rightarrow t^2 - 2t + 1 = 3$
$\Rightarrow (t-1)^2 = 3$
$\Rightarrow t = 1 \pm \sqrt{3}$
$\Rightarrow t = 1 \pm 1.73$
\Rightarrow t = 2.73 or -0.73 (rejected as t > 0)
$\Rightarrow e^{\sin x} = 2.73$
$\Rightarrow \log_e e^{\sin x} = \log_e 2.73$
$\Rightarrow \sin x = \log_e 2.73 > 1$
So no solution.
10. The line passes through the centre of circle $x^2 + y^2 - 16x - 4y = 0$, it interacts with the positive coordinate axis at $A \& B$. Then find the minimum value of $OA + OB$, where O is origin.
(1) 20 (2) 18 (3) 12 (4) 24
(3) 12 (4) 24 Answer (1)
Sol. $(y-2) = m(x-8)$
⇒ x-intercept
$\Rightarrow \left(\frac{-2}{m}+8\right)$
\Rightarrow y-intercept
\Rightarrow (-8 <i>m</i> + 2)
$\Rightarrow OA + OB = \frac{-2}{m^2} + 8 - 8m + 2$
$f'(m)=\frac{2}{m^2}-8=0$
$\Rightarrow m^2 = \frac{1}{4}$

as

(2) $\frac{e^3+2}{\sqrt{e^3+1}}$

(4) e



(Memory Based Solutions)

$$\Rightarrow m = \frac{-1}{2}$$

$$\Rightarrow f\left(\frac{-1}{2}\right) = 18$$

$$\Rightarrow \text{ Minimum = 18}$$
11. If for some $m, n; {}^{6}C_{m} + 2\left({}^{6}C_{m+1}\right) + {}^{6}C_{m+2} > {}^{8}C_{3}$
and ${}^{n-1}P_{3}: {}^{n}P_{4} = 1:8$, then ${}^{n}P_{m+1} + {}^{n+1}C_{m}$ is
equal to
(1) 6756 (2) 7250
(3) 6223 (4) 6550
Answer (1)
Sol. ${}^{6}C_{m} + 2\left({}^{6}C_{m+1}\right) + {}^{6}C_{m+2} > {}^{8}C_{3}$
 ${}^{7}C_{m+1} + {}^{7}C_{m+2} > {}^{8}C_{3}$
 ${}^{8}C_{m+2} > {}^{8}C_{3}$
 $\therefore m = 2$
and ${}^{n-1}P_{3}: {}^{n}P_{4} = 1:8$
 $\frac{(n-1)(n-2)(n-3)}{n(n-1)(n-2)(n-3)} = \frac{1}{8}$
 $\therefore n = 8$
 $\therefore {}^{n}P_{m+1} + {}^{n+1}C_{m} = {}^{8}P_{5} + {}^{9}C_{2}$
 $= 8 \times 7 \times 6 \times 5 \times 4 + \frac{9 \times 8}{2}$
 $= 6756$
12. Let $f: (-\infty, -1] \rightarrow (a, b]$ be defined as $f(x) = e^{x^{3}-3x+1}$, if f is both one and onto, then the distance from a point $P(2a + 4, b + 2)$ to curve

 $:: P(4, e^3 + 2)$ d $x + e^{-3}v = 4$ $d = \frac{(e^3 + 2)(e^{-3})}{\sqrt{1 + e^{-6}}} = \frac{1 + 2e^{-3}}{\sqrt{1 + e^{-6}}} = \frac{e^3 + 2}{\sqrt{e^6 + 1}}$ 13. If (α, β, γ) is mirror image of the point (2, 3, 4) with respect to the line $\frac{x-1}{2} = \frac{y-2}{3} = \frac{z-3}{4}$. Then 2α + $3\beta + 4\gamma$ is (1) 29 (2) 30 (3) 31 (4) 32 Answer (1) A | (2, 3, 4) Sol. Line $B^{(\alpha, \beta, \gamma)}$ Take $\frac{x-1}{2} = \frac{y-2}{3} = \frac{z-3}{4} = \lambda$ $x = 2\lambda + 1$, $y = 3\lambda + 2$, $z = 4\lambda + 3$ $AB = (\alpha - 2)\hat{i} + (\beta - 3)\hat{j} + (\gamma - 4)\hat{k}$ Now, $(\alpha - 2) \cdot 2 + (\beta - 3) \cdot 3 + (\gamma - 4) \cdot 4 = 0$ $2\alpha - 4 + 3\beta - 9 + 4\gamma - 16 = 0$ \Rightarrow 2 α + 3 β + 4 γ = 29 14. A parabola has vertex (2, 3), equation of directrix is 2x - y = 1 and equation of ellipse is $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1, e = \frac{1}{\sqrt{2}}$ and ellipse passing through

 $b = e^{-1+3+1} = e^3 = f(-1)$

focur of parabola then square of length of latus rectum of ellipse is

(1)	6564	(2)	3288 25
(1) $\frac{6564}{25}$	(2)	25	

(3)
$$\frac{6272}{25}$$
 (4) $\frac{4352}{25}$

Answer (3)

Sol.
$$f(x) = e^{x^3 - 3x + 1}$$

 $f'(x) = e^{x^3 - 3x + 1} \cdot (3x^2 - 3)$
 $= e^{x^2 - 3x + 1} \cdot 3(x - 1)(x + 1)$
For $x \in (-\infty, -1], f'(x) \ge 0$
 $\therefore f(x)$ is increasing function
 $\therefore a = e^{-\infty} = 0 = f(-\infty)$

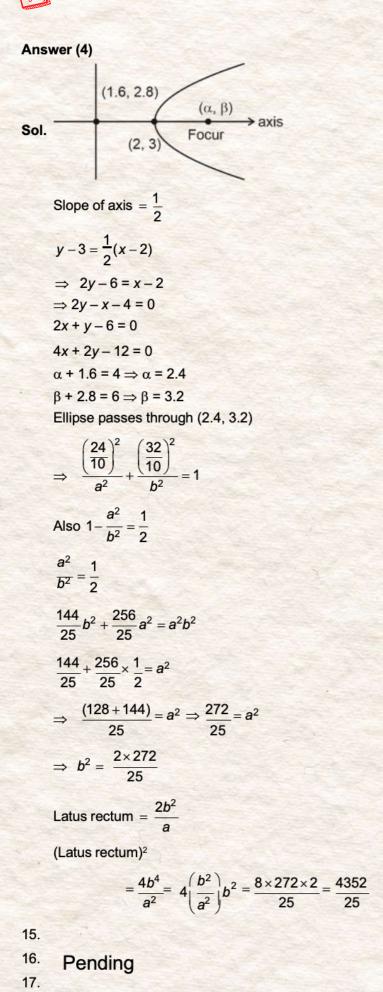
 $x + ye^{-3} - 4 = 0$ is

(1) $\sqrt{e^3+2}$

(3) $\frac{e^3+2}{\sqrt{e^6+1}}$

(Memory Based Solutions)





^{19.} Pending

SECTION - B

Numerical Value Type Questions: This section contains 10 Numerical based questions. The answer to each question should be rounded-off to the nearest integer.

21. The value of
$$\frac{120}{\pi^3} \left| \int_0^{\pi} \frac{x^2 \sin x \cdot \cos x}{(\sin x)^4 + (\cos x)^4} dx \right|$$
 is

Answer (15)

Sol.
$$\int_{0}^{\pi} \frac{x^{2} \sin x \cdot \cos x}{\sin^{4} x + \cos^{4} x} dx$$
$$= \int_{0}^{\frac{\pi}{2}} \frac{\sin x \cos x}{\sin^{4} x + \cos^{4} x} (x^{2} - (\pi - x)^{2}) dx$$

$$\int_{0}^{\frac{\pi}{2}} \frac{\sin x \cdot \cos x (2\pi x - \pi^{2})}{\sin^{4} x + \cos^{4} x} x$$

$$=2\pi\int_{0}^{\frac{\pi}{2}}\frac{x\sin x\cos x}{\sin^{4}x+\cos^{4}x}dx-\pi^{2}\int_{0}^{\frac{\pi}{2}}\frac{\sin x\cos x}{\sin^{4}x+\cos^{4}x}dx$$

$$= 2\pi \cdot \frac{\pi}{4} \int_{0}^{\frac{\pi}{2}} \frac{\sin x \cos x}{\sin^4 x + \cos^4 x} dx - \pi^2 \int_{0}^{\frac{\pi}{2}} \frac{\sin x \cos x}{\sin^4 x + \cos^4 x} dx$$

$$= -\frac{\pi^2}{2} \int_{0}^{\frac{\pi}{2}} \frac{\sin x \cos x}{\sin^4 x + \cos^4 x} dx$$

$$-\frac{\pi^2}{2}\int_{0}^{\overline{j}}\frac{\sin x\cos xdx}{1-2\sin^2 x+\cos^2 x}$$

$$= -\frac{\pi^2}{2} \int_{0}^{\frac{\pi}{2}} \frac{\frac{1}{2} \sin 2x}{1 - \frac{1}{2} \sin^2 2x} dx$$

$$= -\frac{\pi^2}{2} \int_{0}^{\frac{1}{2}} \frac{\sin 2x}{2 - \sin^2 2x} dx$$

$$= -\frac{\pi^2}{2} \int_{0}^{\frac{\pi}{2}} \frac{\sin 2x}{1 + \cos^2 2x} dx$$





Let $\cos 2x = t$

$$= -\frac{\pi^2}{2} \int_{-1}^{1} \frac{-\frac{1}{2} dt}{1+t^2}$$
$$= -\frac{\pi^2}{4} \int_{-1}^{1} \frac{dt}{1+t^2}$$
$$= -\frac{\pi^2}{4} \cdot \frac{\pi}{2} = -\frac{\pi^3}{8}$$
$$\therefore \quad \frac{120}{\pi^3} \left| -\frac{\pi^3}{8} \right| = 15$$

22. The number of ways to distribute the 21 identical apples to three children's so that each child gets at least 2 apples.

Answer (136)

Sol. After giving 2 apples to each child 15 apples left now 15 apples can be distributed in ${}^{15+3-1}C_2 = {}^{17}C_2$ ways

 $=\frac{17\times16}{2}=136$

23. If $A = \{1, 2, 3, ..., 100\}$, $R = \{(x, y) \mid 2x = 3y, x, y \in A\}$ is symmetric relation on A and the number of elements in R is n, the smallest integer value of n is

Answer (0)

Sol. \square *R* is symmetric relation

 $\Rightarrow (y, x) \in R \forall (x, y) \in R$ (x, y) $\in R \Rightarrow 2x = 3y$ and (y, x) $\in R \Rightarrow 3x = 2y$ Which holds only for (0, 0)

Which does not belongs to R.

 \therefore Value of n = 0

24. Matrix A of order 3 × 3 is such that |A| = 2 if $n = |\operatorname{adj}(\operatorname{adj}(\operatorname{adj}...(a)))|$ then remainder when n is 2024 times

20211

divided by 9 is

Answer (7)

Sol. |*A*| = 2

 $adj \left(adj \left(adj \dots \left(a \right) \right) \right) = |A|^{(n-1)^{2024}}$ 2024 times

$$= |A|^{2^{2024}}$$

= $2^{2^{2024}}$

$$2^{2024} = (2^2) 2^{2022} = 4(8)^{674} = 4(9-1)^{674}$$

$$\Rightarrow 2^{2024} \equiv 4 \pmod{9}$$

$$\Rightarrow 2^{2024} \equiv 9m + 4, \quad m \leftarrow \text{even}$$

$$2^{9m+4} \equiv 16 \cdot (2^3)^{3m} \equiv 16 \pmod{9}$$

25. 26.

27. Pending

28. 29.



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